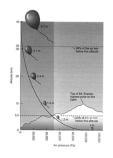
### Lecture 7: Air cools when it rises



- ☐ Air expands as it rises
- ☐ Air cools as it expands
- ☐ Air pressure
- ☐ Lapse rates

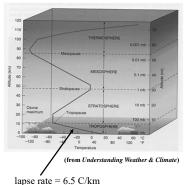
### Why Is the Mountain Peak Cold?

- ☐ Sunlight heats the atmosphere from below
- → Convection occurs and brings heat upward
- → Air parcels expands as they move upward (because air pressure decreases with height in the atmosphere)
- → Thermal energy in the parcel is used to expand the air parcel
- → Air parcel becomes cold
- → Temperature decreases with height in the atmosphere
- → clouds, rains, snows......



### **Vertical Thermal Structure**

### Standard Atmosphere



Troposphere ("overturning" sphere)

- contains 80% of the mass
- · surface heated by solar radiation
- · strong vertical motion
- where most weather events occur

Stratosphere ("layer" sphere)

- weak vertical motions
- dominated by radiative processes
- heated by ozone absorption of solar ultraviolet (UV) radiation
- warmest (coldest) temperatures at summer (winter) pole

Mesosphere

- heated by solar radiation at the base
- heat dispersed upward by vertical motion

Thermosphere

very little mass

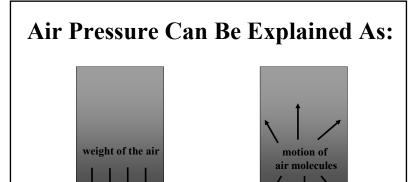


### Three Things Need To Be Explained

- (1) Why air expands as it rises?
- (2) Why air cools as it expands with height?
- (3) What is air pressure?



# 



The weight of air above a surface (due to Earth's gravity)

The bombardment of air mocules on a surface (due to motion)



### **Units of Atmospheric Pressure**

- ☐ Pascal (Pa): a SI (Systeme Internationale) unit for air pressure.

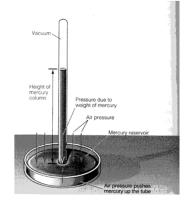
  1 Pa = force of 1 newton acting on a surface of one square meter
  1 hectopascal (hPa) = 1 millibar (mb) [hecto = one hundred = 100]
- ☐ Bar: a more popular unit for air pressure.

 $1 \ bar = 1000 \ hPa = 1000 \ mb$ 

 $\Box$  One atmospheric pressure = standard value of atmospheric pressure at lea level = 1013.25 mb = 1013.25 hPa.



### **One Atmospheric Pressure**

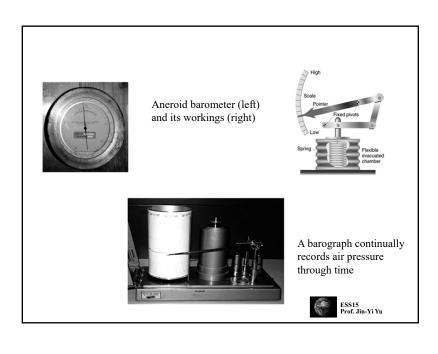


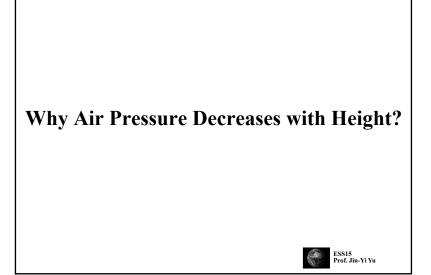
(from The Blue Planet)

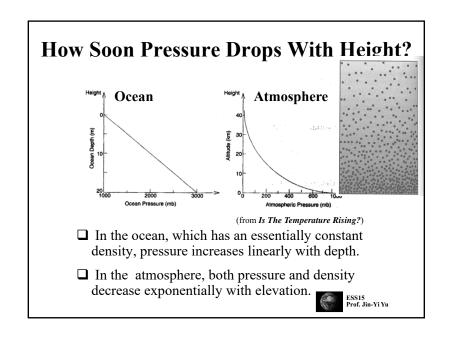
- ☐ The average air pressure at sea level is equivalent to the pressure produced by a column of water about 10 meters (or about 76 cm of mercury column).
- ☐ This standard atmosphere pressure is often expressed as 1013 mb (millibars), which means a pressure of about 1 kilogram per square centimeter.

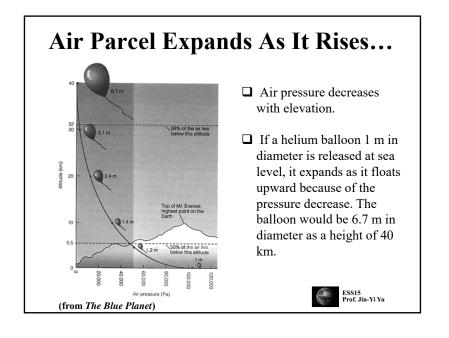


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### What Happens to the Temperature?

- ☐ Air molecules in the parcel (or the balloon) have to use their kinetic energy to expand the parcel/balloon.
- ☐ Therefore, the molecules lost energy and slow down their motions
- → The temperature of the air parcel (or balloon) decreases with elevation.



### The First Law of Thermodynamics

- ☐ This law states that (1) heat is a form of energy that (2) its conversion into other forms of energy is such that total energy is conserved.
- ☐ A parcel of air expands and pushes its surroundings back loses energy in the process.
- ☐ That energy comes from the random motion of the air molecules in the parcel.
- ☐ After the expansion, the molecules move less energetically → the expansion causes the temperature of the air to decrease.



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### **How to Change Air Temperature?**

- ☐ Add (remove) heat to (from) the air parcel (diabatic processes)
  - (1) Conduction: requires touching
  - (2) Convection: Hot air rises
  - (2) Advection: horizontal movement of air
  - (3) Radiation: exchanging heat with space
  - (4) Latent heating: changing the phase of water
- lacktriangle Without adding (removing) heat to (from) the air parcel
  - (1) Adiabatic Process: Expanding and compressing air



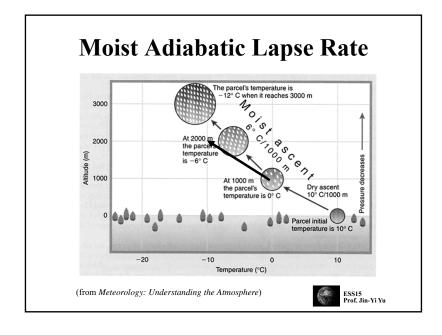
### **Adiabatic Lapse Rate**

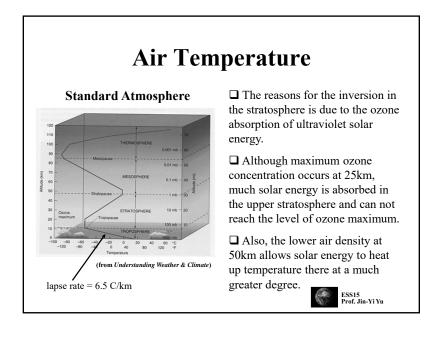
- ☐ By considering the Boyle's law (about random motion of air molecules), the barometric law (about the gravitational pull of air molecules), and the first law of thermodynamics (about the energy conversion), it can be shown that air temperature decrease linearly with height.
- ☐ This linear decreasing rate is called the "adiabatic lapse rate", which is about 10 °C per kilometer.
- ☐ "Adiabatic" means no heat is added or subtracted from a parcel as it rises, expands, and cools.

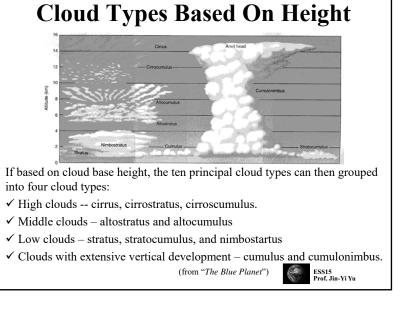


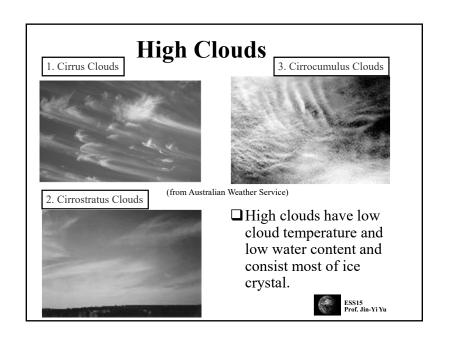
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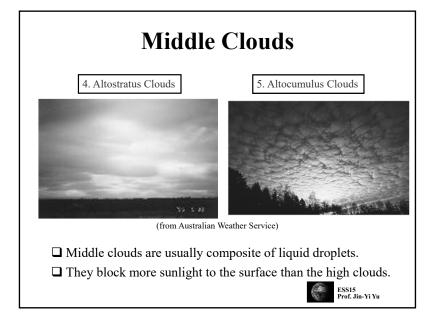
# Dry Adiabatic Lapse Rate 3000 At 3000 m, its temperature is -20° C When it reaches an allitude of 2000 m, its temperature is -10° C After rising to an allitude of 1000 m, its new temperature is 0° C After rising to an allitude of 1000 m, its new temperature is 10° C (from Meteorology: Understanding the Atmosphere) ESS15 Prof. Jin-Yi Vu

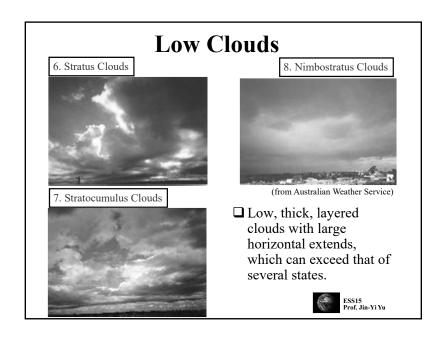


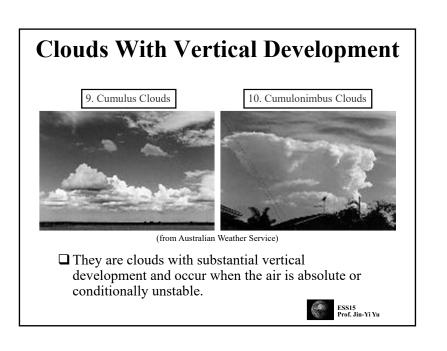












### **Forms of Precipitation**

- $\Box$ Rain
- $\square$  Snow
- ☐ Graupel and Hail
- $\Box$  Sleet
- □ Freezing Rain



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### Rain









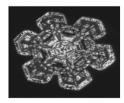
- Rain is associated with warm clouds exclusively and cool clouds when surface temperatures are above freezing
- Rainshowers are episodic precipitation events associated with convective activity and cumulus clouds
  - Drops tend to be large and widely spaced to begin, then smaller drops become more prolific
- Raindrop Shape begins as spherical
  - · As frictional drag increases, changes to a mushroom shape
  - · Drops eventually flatten
  - Drops split when frictional drag overcomes the surface tension of water
  - Splitting ensures a maximum drop size of about 5 mm and the continuation of the collision-coalescence process



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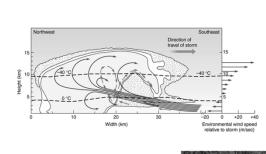
Hail Formation

### Snow



- ☐ Snowflakes have a wide assortment of shapes and sizes depending on moisture content and temperature of the air.
- ☐ Snowfall distribution in North America is related to north-south alignment of mountain ranges and the presence of the Great Lakes.
- ☐ Lake effect: snows develop as the warm lake waters evaporate into cold air.



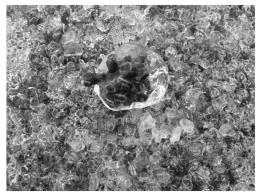


Concentric layers of ice in hail indicate the cyclical hailstone formation process



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## **Freezing Rain and Sleet**



(Photographer: Lee Anne Willson)

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