

## Lecture 11: Ozone Hole Depletion



- Ozone Production and Destruction
- Catalytic Cycle
- Polar Stratospheric Clouds
- CFC and Ozone Depletion

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## Ozone Production and Destruction

(from *The Earth System*)

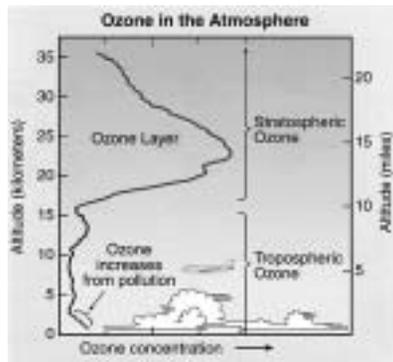
| The Chapman Mechanism of Ozone Production and Destruction |  |      |
|---|--|------|
| Reaction*   |  | Rate |
| 1)  | $O_2 + \text{UV photon} \rightarrow O + O$ | Slow |
| 2)  | $O + O_2 + M \rightarrow O_3 + M$          | Fast |
| 3)  | $O_3 + \text{photon} \rightarrow O_2 + O$  | Fast |
| 4)  | $O + O_3 \rightarrow 2 O_2$                | Slow |

visible light

destroy O3 permanently

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## Ozone Distribution



(from WMO Report 2003)

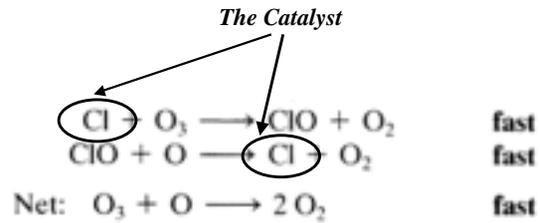
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## Other Ozone Destruction Processes

- Other atmospheric trace constituents, such as nitrous oxide (N<sub>2</sub>O), water vapor, and freons, can also be photolyzed. They produce highly reactive radicals that keep ozone abundances lower than they would otherwise be.
- These radicals include: nitric oxide (NO), atomic chlorine (Cl), bromine (Br) radicals, and hydroxyl (OH) radicals.
- These radicals can destroy stratospheric ozone through “catalytic cycles”.

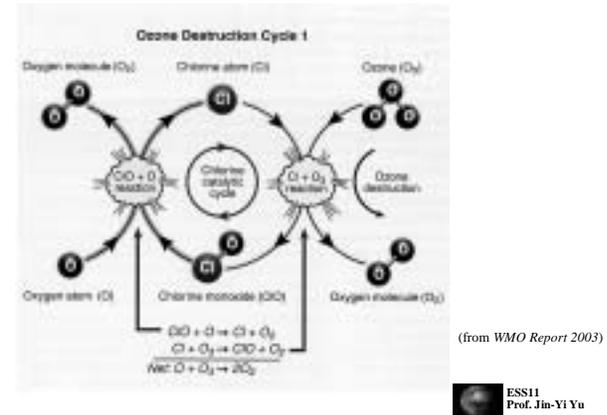
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## The Chlorine Catalytic Cycle



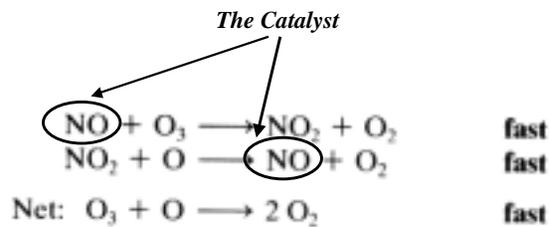
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## The Chlorine Catalytic Cycle



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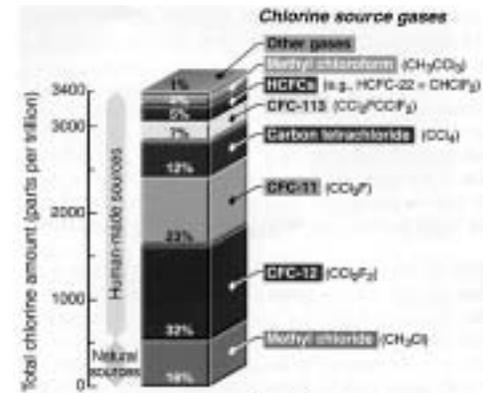
## The Nitrogen Catalytic Cycle



(from The Earth System)

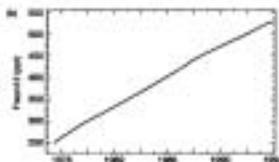
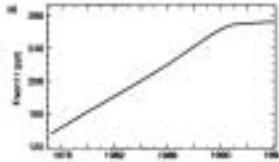
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## Chlorine Sources



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## Man-Made Sources for CFCs



(from *The Earth System*)

- ❑ There are two kinds of CFCs: freon-11 (CCl<sub>3</sub>F) and freon-12 (CCl<sub>2</sub>F<sub>2</sub>).
- ❑ Freon-11 has been used:
  - (1) as a propellant in spray cans
  - (2) as a blowing agent for producing foams
  - (3) to clean semiconductor chips.
- ❑ Freon-12 has been used as
  - (1) a refrigerant
  - (2) working fluid in most car air conditioners.

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## Polar Stratospheric Clouds (PSCs)



(Sweden, January 2000; from NASA website)

- ❑ In winter the polar stratosphere is so cold ( $-80^{\circ}\text{C}$  or below) that certain trace atmospheric constituents can condense.
- ❑ These clouds are called “polar stratospheric clouds” (PSCs).

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## How PSCs Affect Ozone Hole

- ❑ The ice crystals in the polar stratospheric clouds provide surface for the ozone depletion surface to occur more easily.
- ❑ On these cloud surfaces, certain forms of chlorine that do not react with ozone are converted into forms that do.
- ❑ Polar stratospheric clouds set up the stage for massive destruction of ozone to happen when sunlight returns in the spring.

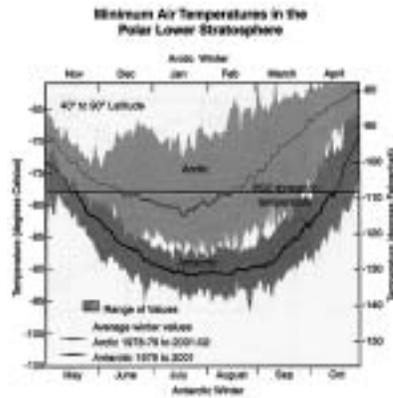
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## Two Factors for the Ozone Hole

- ❑ Very Cold temperature
  - To form polar stratospheric clouds (PSCs)
- ❑ Return of the Sunlight
  - To start the ozone depletion reaction

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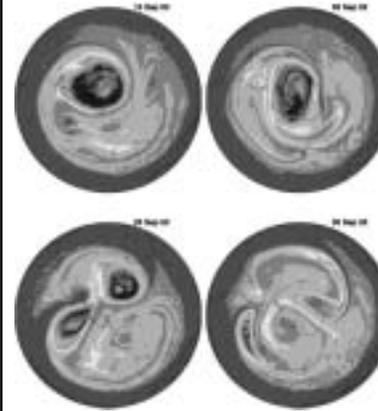
## Why No Ozone Hole in Arctic?



(from WMO Report 2003)

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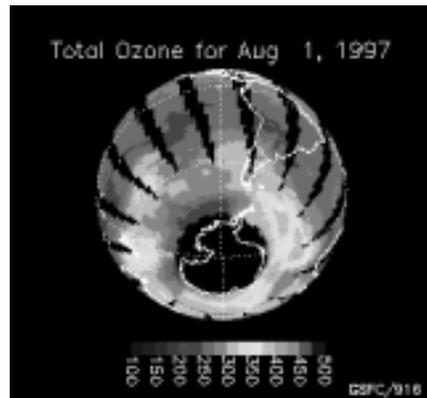
## The Polar Vortex



- ❑ The wintertime circulation over the South Pole is characterized by a gigantic whirlpool of cold and dense air, called the polar vortex.
- ❑ The cold and dense cold air in the middle of the vortex is subsiding.
- ❑ The sinking air carries cloud particles along with it.
- ❑ Remove odd nitrogen from the stratosphere.
- ❑ Very little ozone and odd nitrogen can be brought into the south pole.

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## The 1997 Ozone Hole



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